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## The efficacy of *Ximenia americana* plant mediated silver nanoparticles against dengue vector mosquito larvae [*Aedes (Stegomyia) aegypti* (Linnaeus, 1762) (Diptera: Culicidae)]

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### Abstract

Nanotechnology is an emerging technology and day to day expanding its roots of knowledge to various branches of Science and fulfilling human conveniences. To manage vector populations, new tools have been emerging through nanotechnology. Nanoparticles possess peculiar toxicity mechanisms due to surface volume ratio, and this may actively contribute to their excellent larvicidal potential against *Aedes aegypti* mosquito. Dengue is a mosquito borne disease and making major public health issue worldwide. *Aedes (Stegomyia) aegypti* (Linnaeus, 1762) prefers to breed in artificial containers, and anti-larval measures with *Ximenia americana* L. (Olacaceae) leaf mediated silver nanoparticles (AgNPs) proves to be a potential substitute for the existing organophosphorus insecticides like temephos, malathion and fenthion etc., for mosquito control programme. Larvae were exposed to varying concentrations of plant extracts and synthesized silver nanoparticles for 24 hours. From the results, it was found that plant extracts showed moderate larvicidal effects (LC50 at 179.87 and LC90 at 376.77 ppm) but, the synthesized silver nanoparticles had found to be toxic to larvae at LC50 (0.63 ppm) and LC90 (1.20 ppm). This research highlighted that, the *Ximenia americana* leaf mediated AgNPs are an efficient and eco-friendly agents against *Ae. aegypti* mosquito, though the laboratory studies have shown the promising results, yet their efficacy in the field is to be tested for effective mosquito larval control.

**Keywords:** *Aedes aegypti*, *Ximenia americana* L. (Olacaceae), Silver nanoparticles (AgNPs)

### 1. Introduction

The incidence of dengue has grown dramatically around the world in recent decades and making major public health concern worldwide [1-6]. One recent estimate indicates 390 million dengue infections per year (95% credible interval 284–528 million), of which 96 million (67–136 million) manifest clinically (with any severity of disease) [7, 8]. Another study on the prevalence of dengue estimates that 3.9 billion people in 128 countries are at risk of infection with dengue viruses [9]. Dengue is widespread throughout the tropics with local variations in risk influenced by temperature, rainfall and unplanned rapid urbanization. In India, a total of 101192 cases and 172 deaths were recorded due to dengue for the year of 2018 and in 2019 it was increased to 136422 whereas the death rate was decreased to 132 [10].

*Aedes (Stegomyia) aegypti* (Linnaeus, 1762) is a day-time feeder and prefers to breed in manmade containers and thrives in urban and peridomestic environments where it transmits the Dengue virus to humans [11]. This mosquito also transmits Chikungunya, Yellow fever and Zika infection. Long time exposure to various synthetic insecticides leads to the occurrence of the resistance in vector mosquitoes [12]. Acquired insecticide resistance has reduced the ability of insecticides to control mosquito vectors [13-15]. Synthetic insecticides had unique properties, such as long time persistence, more residual activity and biomagnifications. These properties could be hazardous and lead to the chronic effects in non-target organisms, and the environment [16-18]. Plant mediated insecticides are an outstanding alternative for the synthetic insecticides [19]. Number of phytochemicals with biological activity against larval and adult mosquitoes has been described [20-24].

Involvement of nanotechnology in arthropod vector control led to emergence of new routes such as green synthesis of metallic nanoparticles against mosquito and it has been succeeded

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and giving the best solutions to mosquito menace [25-27]. The plant-mediated synthesis of silver nanoparticles; is an eco-friendly approach, more advantageous over chemical and physical methods, cheap, single-step and does not require high pressure, energy, temperature, or extremely toxic chemicals [28, 29]. *Ximenia americana* belongs to the family Olacaceae, commonly known as sea lemon. It is a bush forming shrub and small tree and found on banks of rivers and moist areas throughout India. The aim of this study is to determine the biological activity of *Ximenia americana* leaf extract against *Ae. aegypti* mosquitoes. In the Laboratory the efficacy of leaf extract was tested with crude as well as plant mediated silver nanoparticles with 3<sup>rd</sup> and 4<sup>th</sup> instar larvae.

## 2. Materials and Methods

### 2.1 Mosquito collection and culture

*Ae. aegypti* larvae and pupae were collected by the dipping method from natural habitats, in and around Osmania University, Hyderabad, Telangana, India [30]. Rearing conditions for all mosquitoes were maintained (27±2 °C, 75–85% RH, and L14: D10 photoperiod). According to the manual for mosquito rearing and experimental techniques published by American Mosquito Control Association larvae were reared to adulthood in the mosquito culture room in two liter of water in white enamel trays (30 cm long×25 cm wide×6 cm deep) where they were provided a daily food mix comprising three parts brewer's yeast and one part dog biscuit [31]. Pupae were placed into screened cages (23 cm long×23 cm wide×23 cm deep). When imago emerged after 24 hours; they were identified followed by standard keys [32-34], transferred into glass cages (30 cm long×30 cm wide×30 cm deep) and provided with 10% sucrose solution (in water) via a piece of cotton. After mating, female mosquitoes were allowed access to feed on nude mice for blood feeding. Eggs were collected through ovitraps and this pure culture was reared again to get the F<sub>1</sub> generation and finally these larvae were used for bioassay.

### 2.2 Plant material collection and preparation of extract:

Plant material collection and extraction has been done by standard methodologies [35, 36]. *Ximenia americana* plant leaves were collected from Osmania University, Hyderabad, Telangana, India (Latitude: 17.427 and Longitude: 78.53448). Taxonomic identification carried out at the Department of Botany, Osmania University, Hyderabad, India. A voucher specimen was deposited in the Medical Entomology Laboratory (ID: NHMZD273), Department of Zoology, Osmania University, Hyderabad.

Fresh leaves of *Ximenia americana* were collected, washed several times with tap water to remove the dust particles and then shade dried for two weeks at 27 °C to remove the residual moisture and grinded to form fine powder. Then plant extract was prepared by mixing 25 g of leaf powder with 250 mL of deionized water in a 500 mL of (Borosil, India) conical flask. The solution was mixed for every 3 hours with fresh glass rod. After 48 hours the solvent color changed from moderate to fully dark; it indicates dissolved state of plant material. Plant material was filtered with the help of What's man No. 1 filter paper, and crude extract was stored at 4 °C for further analysis. Presence of alkaloids was confirmed using Mayer's test and Wagner's test. Then the solution was used for the reduction of silver ions (Ag<sup>+</sup>) to silver nanoparticles (Ag<sup>0</sup>).

### 2.3 Synthesis of Silver Nanoparticles

Biosynthesis of silver nanoparticles has been carried out by standard methodologies [37-40]. In a typical synthesis of silver (Ag) nanoparticles, 20 ml of leaf extract was added to reduce 80 ml of 1 mM AgNO<sub>3</sub> (bought from Sigma Aldrich Bangalore) aqueous solution and kept at room temperature. For the reproducibility experiment was done in triplicate.

### 2.4 Characterization of AgNPs

The bioreduction of pure Ag<sup>+</sup> ions was monitored by measuring the UV-visible spectra (UV-spec) of the reaction medium [Shimadzu 2600 – (TCC)]. For FTIR measurements, the Silver nanoparticles solution was centrifuged at 10,000 rpm for 30 min. The pellet was washed three times with 20 ml of deionized water to get rid of the free proteins or enzymes that are not capping the silver nanoparticles. The samples were dried and grinded with KBr pellets and analysed on a Bruker Optics (Germany made) Tensor-27 model in the diffuse reflectance mode operating at a resolution of 0.4 cm<sup>-1</sup>. The size and shape of silver nanoparticles have been visualized through a scanning electron microscope (SEM). Presence of elemental study was carried out by using EDS. Crystalline nature of AgNPs was studied through X-ray diffraction (XRD). GC-MS analysis has been carried out by the SHIMADZU QP2010, an oven temperature from 50 to 280 °C at 4°C/min and held at this temperature for 5 min; inlet and interface temperatures were 250 °C and 280 °C, respectively. Carrier gas was He at a flow rate of 1.0 ml/min (constant flow). 0.2 ml of sample was injected under a split of 20:1. EIMS: electron energy, 70 eV. Interpretation of mass spectrum GC-MS was conducted using a database of NIST, having more than 62,000 patterns. The spectrum of the known compounds was compared with the NIST library.

### 2.5 Bioassay

Bioassay test was performed according to WHO guidelines [41], with different concentrations to assess the larvicidal activity. Percentage mortality was calculated as follows:

$$\text{Percentage mortality} = (\text{Number of dead individuals} / \text{Number of treated individuals}) \times 100$$

### 2.6 Statistical analysis

The average larval mortality data were subjected to Probit analysis (FORTRAN) for calculating LC<sub>50</sub> and LC<sub>90</sub> [42].

## 3. Results

After 1 hour the color of the solution changed from colorless to honey brown (Figure-1) indicating the formation of silver nanoparticles and this was confirmed by UV-visible spectroscopy. At 60 minutes a peak value at 435 nm with intensity of 0.06 was observed corresponding to AgNPs (Figure-2). Figure-3 illustrates the X-ray diffraction pattern with three characteristics of diffraction peaks for the AgNPs synthesized from the aqueous leaf extract of *Ximenia americana*. Bragg reflections with 2θ values of 40.63°, 45.74°, and 77.50° were indexed for the planes (111), (200) and (311) respectively. These Bragg refractions were in agreement with reference pattern of JCPDS NO.04-0783 and thus conforms the synthesized silver nanoparticle, which were crystalline in nature (size 27.01 nm) and of face centered cubic (FCC) crystal lattice (Figure-3).

Infra-red (IR) spectra of AgNPs synthesized using *Ximenia americana* aqueous leaf filtrate after bio reduction of silver obtained by FTIR were recorded between 400 and 4000 cm<sup>-1</sup> (Figure-4). Troughs were observed at 3444.98 cm<sup>-1</sup>, 3421.83

$\text{cm}^{-1}$ ,  $2922.25 \text{ cm}^{-1}$ ,  $2854.74 \text{ cm}^{-1}$ ,  $2333.94 \text{ cm}^{-1}$  and  $1629.90 \text{ cm}^{-1}$ . The intense band at  $3444.98 \text{ cm}^{-1}$  is corresponds to O-H stretching,  $3421.83 \text{ cm}^{-1}$  is O-H Medium stretching,  $2922.25 \text{ cm}^{-1}$  asymmetric stretching vibration of  $\text{CH}_2$  of acyl chains (lipids),  $2854.74 \text{ cm}^{-1}$  is corresponding to  $\text{CH}_2$  symmetric stretching,  $2333.94 \text{ cm}^{-1}$  is O-H stretching, the band at  $1629.90 \text{ cm}^{-1}$  is corresponds to amide-I. It is observed from the spectra of silver nanoparticles the appeared bands at  $1629.90$  and  $3444.98$  which are due to amide-I and hydroxyl group that are responsible for reducing the  $\text{Ag}^+$  ions to atoms (Figure-4).

According to the scanning electron micrograph, the morphology of the AgNPs was observed and approximately spherical. In the analysis by energy dispersive spectroscopy (EDS) of the AgNPs the presence of elemental metal signal was confirmed (Figure-5A and B). AFM data analyzed by NOVA-TX software highlighted that *Ximenia americana* leaf mediated AgNPs has shown a size ranging from 10 to 90 nm, with most of them falling within range 10 to 90 nm (Figure-6A and B). From the GC-MS analysis the active ingredient in the synthesized silver nanoparticle responsible for larvicidal activity was found to be 2-propanone (CAS) Acetone \$\$ Propa (Figure-7).  $\text{LC}_{50}$  and  $\text{LC}_{90}$  were recorded at 179.87 and 376.77 ppm with *Ximenia americana*, whereas synthesized silver nanoparticles had shown  $\text{LC}_{50}$  at 0.63 and  $\text{LC}_{90}$  at 1.20 ppm (Table-1, 2 and Figure-8, 9).

#### 4. Discussion

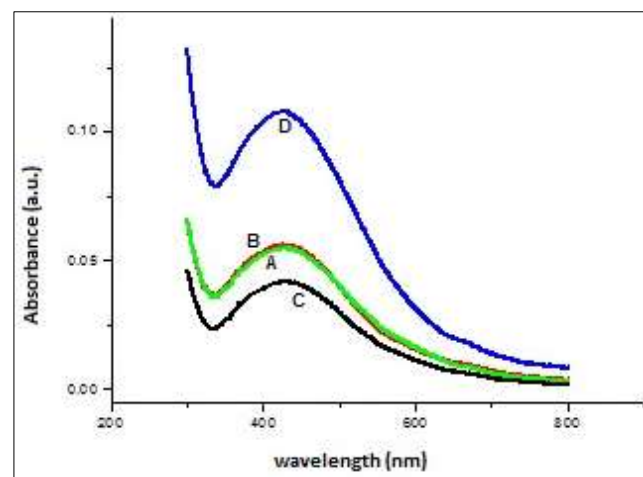
The biofabrication of the mosquitocidal silver nanoparticles is rapid, eco-friendly and cost-effective. AgNPs have excellent anti-mosquito larvicidal activity against third instar larvae of *Ae. aegypti*. Phytochemicals are botanicals which are naturally occurring insecticides obtained from floral resources. Applications of phytochemicals in mosquito control have been in use since the 1920's [43], but the discovery of synthetic insecticides such as DDT in 1939 side tracked the application of phytochemicals in mosquito control programmes. After facing several problems due to injudicious and over application of synthetic insecticides in nature, re-focus on phytochemicals that are easily biodegradable and have no ill-effects on non-target organisms was appreciated. Safe and efficacious insecticides of plant origin have gained importance in recent years and are considered as less hazardous to human health. Studies on the larvicidal action of terrestrial plant extracts against the mosquito larvae were carried out tremendously. Many authors have studied the larvicidal action of terrestrial plant extracts against different mosquito larvae at very high concentrations of the plant extracts for achieving significant mortality of mosquito larvae [44].

Phytochemicals being derived from plant sources can act as larvicide, insect growth regulators, repellent and oviposition attractant and have different activities observed by many researchers. However, insecticides of plant origin have been extensively used on agricultural pests and to a very limited extent, against insect vectors of public health importance. From the results, it was found that plant crude extract showed moderate ( $\text{LC}_{50}$  at 179.87 and  $\text{LC}_{90}$  at 376.77ppm) larvicidal effect but, the synthesized silver nanoparticles had found to be toxic against mosquito larvae at  $\text{LC}_{50}$  (0.63ppm) and  $\text{LC}_{90}$  (1.20ppm). The research findings are in alignment with previously reported research studies. Santhosh kumar *et al.*, (2011) synthesized AgNPs using aqueous crude extract of *Nelumbo nucifera* and tested its efficacy against *Cx. quinquefasciatus*.  $\text{LC}_{50}$  and  $\text{LC}_{90}$  values were recorded at

0.69 and 3.59 ppm [45]. *Turbinaria ornate* mediated AgNPs were synthesized, used against *Ae. aegypti* and  $\text{LC}_{50}$  was recorded at  $0.738 \mu\text{g} / \text{mL}$  [46]. Azarudeen *et al.*, (2017) synthesized AgNPs using *Merremia emarginata*, and got  $\text{LC}_{50}$  at  $9.20 \mu\text{g}/\text{mL}$  against *Ae. aegypti* and  $10.02 \mu\text{g}/\text{mL}$  for *Cx. quinquefasciatus* [47]. Govindarajan *et al.*, (2017) used *Hugonia mystax* to synthesize the AgNPs and got  $\text{LC}_{50}$  at  $15.56 \mu\text{g}/\text{mL}$  for *Ae. aegypti* and  $17.46 \mu\text{g}/\text{mL}$  for *Cx. quinquefasciatus* [48]. *Curcuma zedoaria* mediated AgNPs were tested against *Cx. quinquefasciatus* and  $\text{LC}_{50}$  was recorded at 0.64ppm [49]. Aina *et al.*, (2019) tested the toxicity of *Chasmanthera dependens* mediated AgNPs against *Ae. aegypti* and  $\text{LC}_{50}$  was recorded at  $7.15 \mu\text{g}/\text{mL}$  [50]. Benelli *et al.*, (2018a) synthesized AgNPs using *Acacia caesia* and tested the larvicidal toxicity against *Ae. albopictus* and got  $\text{LC}_{50}$  value at  $11.32 \mu\text{g}/\text{mL}$  [51]. Muthukumaran *et al.*, (2015) concluded that the aqueous crude extracts and synthesized silver nanoparticles of *Gmelina asiatica* had shown the toxicidal effect on late 3<sup>rd</sup> instar of *Cx. quinquefasciatus* at  $\text{LC}_{50}$  value of 139.17mg/l and  $\text{LC}_{90}$  value was 243.39mg/l [52]. Mondal *et al.*, (2019) synthesize silver nanoparticles with *Colocasia esculenta* as a reducing agent and to evaluate their effect against *Culex quinquefasciatus* the  $\text{LC}_{50}$  and  $\text{LC}_{90}$  synthesized silver nanoparticles were 5.17 mg/L and 17.32 mg/L [53].



**Fig 1:** Showing the photograph (A—leaf extract, B—silver nanoparticles after 1 hour) of leaf extract and AgNPs of the *Ximenia americana*.



**Fig 2:** Showing the UV-visible spectra recorded as a function of time of reaction of an aqueous solution of 1 mM  $\text{AgNO}_3$  with the leaf extract of *Ximenia americana* (60 minutes). (a) 30 minutes; (b) 60 minutes; (c) 90 minutes; (d) 120 minutes.

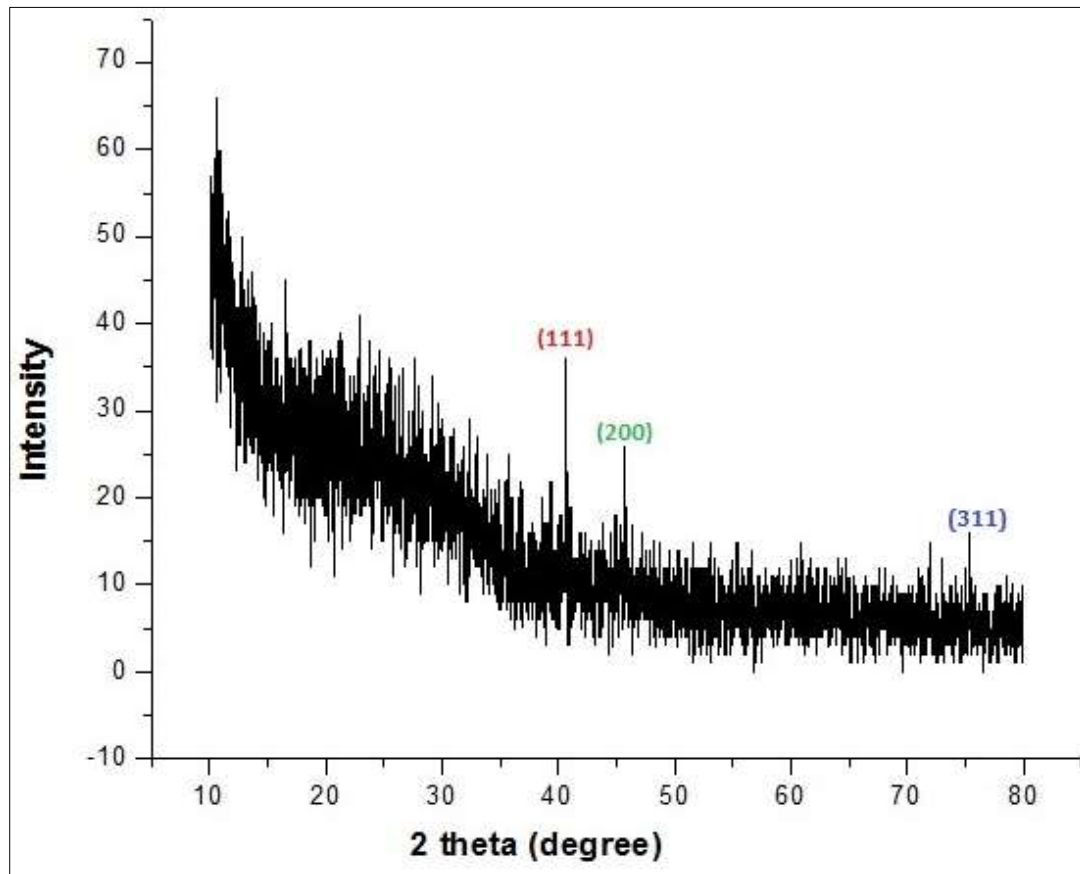


Fig 3: Showing the XRD patterns of synthesized silver nanoparticles of *Ximenia americana*.

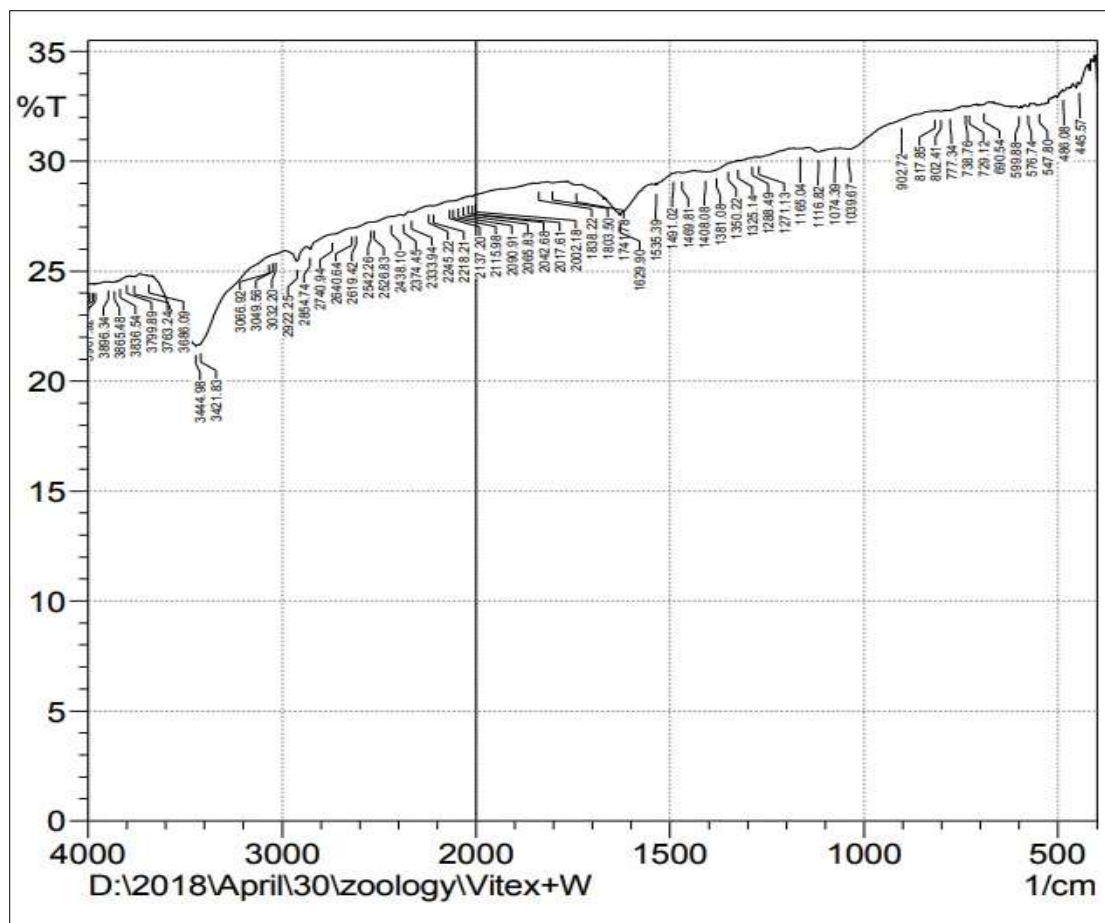
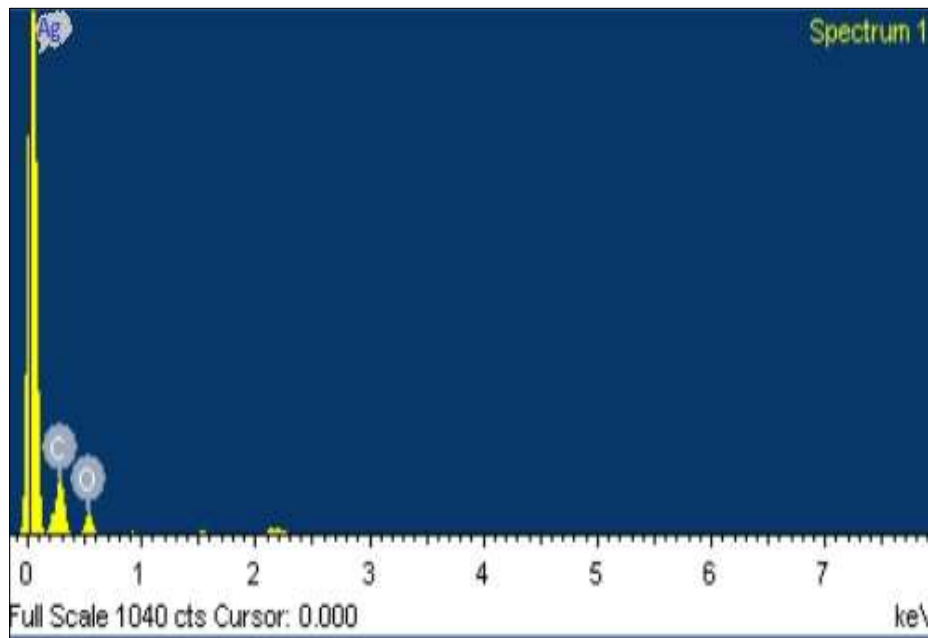
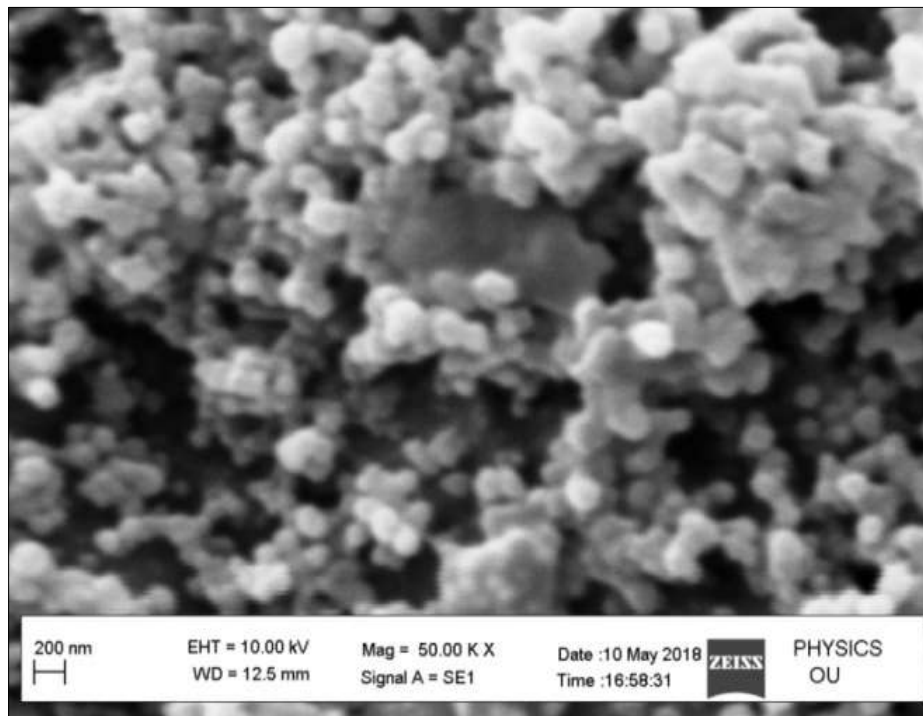
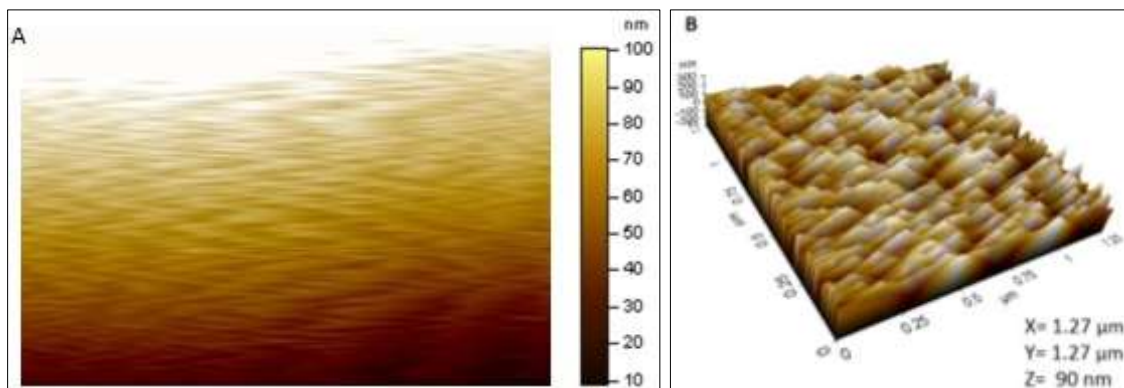


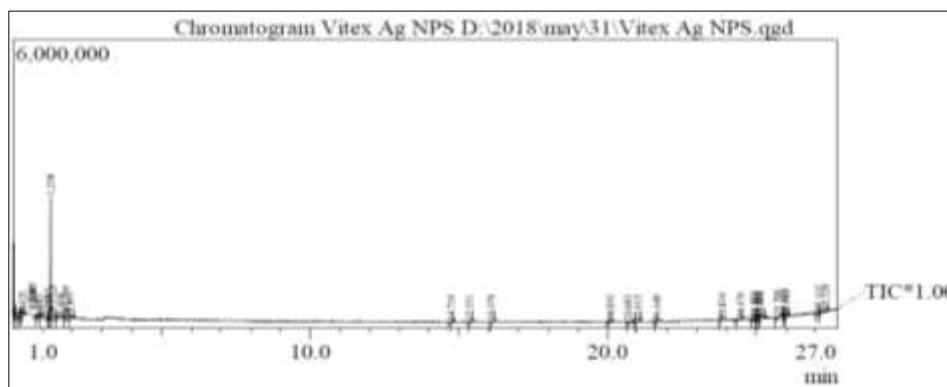
Fig 4: Showing the FT-IR spectrum of AgNPs prepared from the *Ximenia americana* plant extract.



**Fig 5:** Showing the the SEM image (A) and EDS profile (B) of bio-reduced silver nanoparticles.



**Fig 6:** (A) 2.5  $\mu\text{m}$  resolution studies of 10-90 nm size, sperical shaped, polydispersed particles, (B) 3D image of silver nanoparticles analyzes by NOVA-TX software.



Peak Report TIC

| pk# | R.Time | Area    | Area% | Height  | Base m/z | Name  |
|-----|--------|---------|-------|---------|----------|---|
| 1   | 0.045  | 376698  | 4.30  | 152041  | 44.00    | 1-Gala-1-ido-octose \$\$                    |
| 2   | 0.208  | 568588  | 6.49  | 124435  | 44.00    | 1,3,5-Triazin-2(1H)-one, 4,6-bis(ethyl      |
| 3   | 0.240  | 333101  | 3.80  | 125564  | 40.00    | Piperidine, 1-nitro- (CAS) N-Nitropip       |
| 4   | 0.325  | 168757  | 1.93  | 41440   | 42.05    | .ALPHA.-D4-HEXAMETHYLENE O                  |
| 5   | 0.783  | 43664   | 0.50  | 15089   | 44.00    | Benzeneethanamine, .alpha.-methyl-          |
| 6   | 0.883  | 75199   | 0.86  | 15505   | 40.00    | Ethanol, 2,2-dichloro- (CAS) 2,2-Dich       |
| 7   | 1.167  | 343932  | 3.93  | 159836  | 40.00    | .BETA.-IONONE EPOXIDE \$\$                  |
| 8   | 1.258  | 2321655 | 26.51 | 2527057 | 43.00    | 2-Propanone (CAS) Acetone \$\$ propa        |
| 9   | 1.325  | 991120  | 11.32 | 249027  | 44.00    | (S)-(+)-1-Cyclohexylethylamine \$\$         |
| 10  | 1.600  | 595666  | 6.80  | 57494   | 43.05    | Acetic acid (CAS) Ethylic acid \$\$ Vir     |
| 11  | 1.769  | 1377653 | 15.73 | 217283  | 45.05    | 3-hydroxy-2-butanone \$\$ ACETOIN \$        |
| 12  | 1.891  | 292524  | 3.34  | 89081   | 45.05    | 2-Butanone, 3-hydroxy- (CAS) Acetoi         |
| 13  | 14.716 | 30432   | 0.35  | 12834   | 39.95    | 2-Pyrrolidinethione (CAS) Thiopyrrol        |
| 14  | 15.351 | 50298   | 0.57  | 14692   | 44.00    | 1,1-Cyclopropanedicarboxamide \$\$          |
| 15  | 16.074 | 70683   | 0.81  | 18389   | 44.00    | 6-Dimethyl(trimethylsilyl)silyloxytetra     |
| 16  | 20.092 | 53522   | 0.61  | 16024   | 44.05    | Cystine \$\$ L-Cystine \$\$ .beta...beta.'I |
| 17  | 20.683 | 100461  | 1.15  | 15272   | 44.00    | Imidazole, 2-amino-5-[(2-carboxy)vin        |
| 18  | 21.015 | 102243  | 1.17  | 14927   | 44.00    | 5-Nitro-3-nitrylpyridone-2-(1H) \$\$        |

| pk# | R.Time | Area    | Area%  | Height  | Base m/z | Name                                    |
|-----|--------|---------|--------|---------|----------|---|
| 19  | 21.649 | 32490   | 0.37   | 17422   | 44.00    | Benzeneethanamine, 4-chloro-.alpha.-    |
| 20  | 23.834 | 33384   | 0.38   | 14998   | 44.05    | 2-Propanamine, 1-methoxy-               |
| 21  | 24.470 | 58685   | 0.67   | 13472   | 40.00    | 4-METHYL-5-VINYL THIAZOLE \$\$          |
| 22  | 24.900 | 59298   | 0.68   | 16955   | 281.00   | ACETYL-ISO-CODEINE \$\$                 |
| 23  | 25.043 | 58390   | 0.67   | 21221   | 208.05   | Benzo[b]thiophene, 7-ethyl- (CAS)       |
| 24  | 25.075 | 21688   | 0.25   | 18040   | 207.00   | Brallobarbitol \$\$ 2,4,6-(1H,3H,5H)-Py |
| 25  | 25.108 | 105943  | 1.21   | 22346   | 207.00   | 5-Bromo-8-(5-nitrosalicylideneamino)    |
| 26  | 25.708 | 169957  | 1.94   | 20749   | 44.05    | .alpha.-D-Galactopyranose, 2-(acetyla   |
| 27  | 25.950 | 24049   | 0.27   | 16388   | 44.00    | Thiodipropionic amide \$\$              |
| 28  | 25.983 | 58208   | 0.66   | 15619   | 73.00    | 9,12,15-Octadecatrienoic acid, 2-[(tri  |
| 29  | 27.102 | 53229   | 0.61   | 16094   | 206.95   | 6-Methyl-5-[1-piperidinyl]-2,4-pyrimi   |
| 30  | 27.339 | 185802  | 2.12   | 24057   | 207.00   | 2-(N-ethylimino)-3,3-dimethyl-1-ethyl   |
|     |        | 8757319 | 100.00 | 4083351 |          |   |

Fig 7: Showing the GC-MS Analysis of *Ximenia americana* leaf extract.

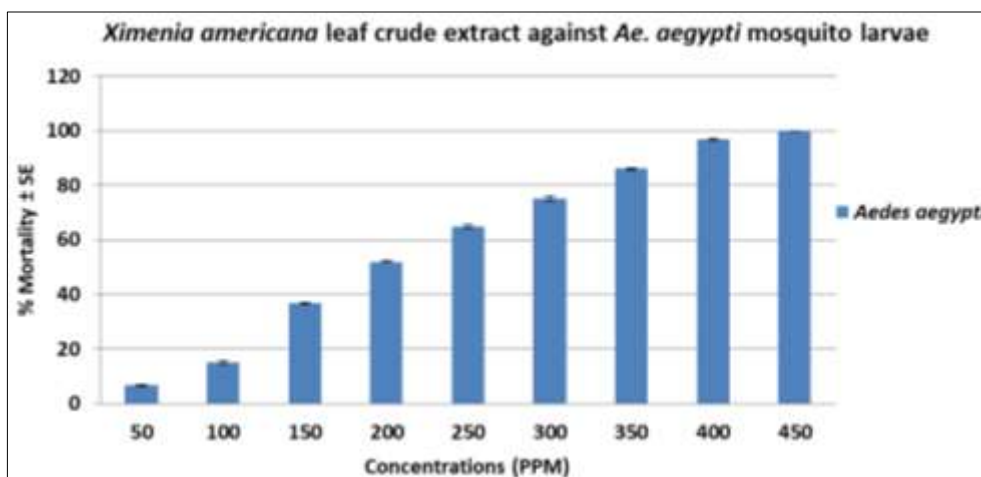


Fig 8: Showing the efficacy of *Ximenia americana* leaf extract against *Ae. aegypti* larvae.

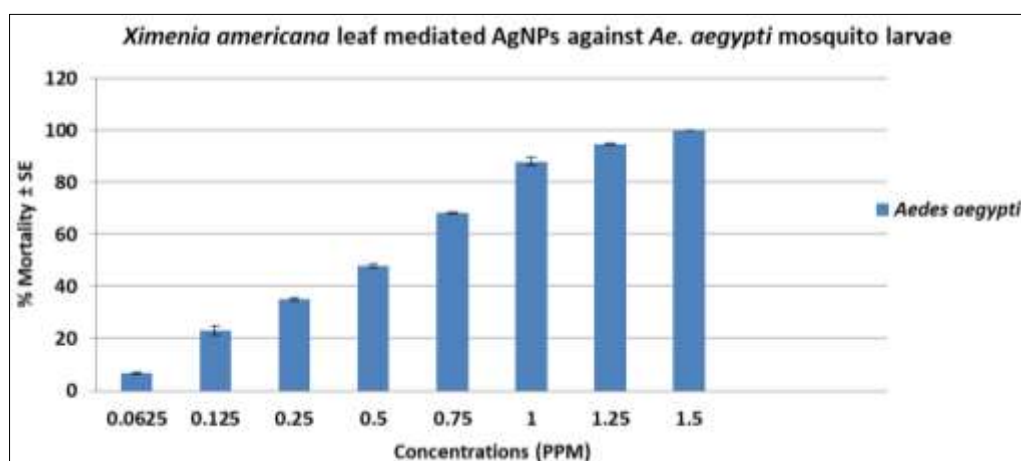


Fig 9: Showing the efficacy tested with *Ximenia americana* leaf mediated silver nanoparticles against *Ae. aegypti*.

Table 1: Showing the efficacy of *Ximenia americana* leaf extract against *Ae. aegypti* larvae.

| Extract             | Concentrations (in PPM) | Mortality (%) ± SE | Lethal concentrations                          |
|---------------------|-------------------------|--------------------|--|
| Control             | 0                       | 0                  |  |
| Plant crude extract | 50                      | 6.67±0.33          | LC50 (PPM) (LCL-UCL)<br>179.87 (165.23-194.26) |
|                     | 100                     | 15±0.57            |  |
|                     | 150                     | 36.67±0.33         |  |
|                     | 200                     | 52±0.57            |  |
|                     | 250                     | 65±0.57            | LC90 (PPM) (LCL-UCL)<br>376.77 (340.22-428.03) |
|                     | 300                     | 75±1               |  |
|                     | 350                     | 86±0.57            |  |
|                     | 400                     | 96.67±0.33         |  |
| 450                 | 100                     |                    |  |

Table 2: Showing the efficacy of *Ximenia americana* leaf extract mediated silver nanoparticles against *Ae. aegypti* larvae.

| Extract           | Concentrations (in PPM) | Mortality (%) ± SE | Lethal concentrations                    |
|-------------------|-------------------------|--------------------|--|
| Control           | 0                       | 0                  |  |
| Synthesized AgNPs | 0.0625                  | 6.67±0.33          | LC50 (PPM) (LCL-UCL)<br>0.63 (0.50-0.78) |
|                   | 0.125                   | 23.33±1.76         |  |
|                   | 0.25                    | 35±0.57            |  |
|                   | 0.5                     | 48±0.57            |  |
|                   | 0.75                    | 68.33±0.33         | LC90 (PPM) (LCL-UCL)<br>1.20 (0.94-1.68) |
|                   | 1                       | 88±1.52            |  |
|                   | 1.25                    | 94.67±0.33         |  |
|                   | 1.5                     | 100                |  |

## 5. Conclusion

Use of these botanical derivatives in mosquito control instead of synthetic insecticides could reduce the cost and environmental pollution. *Ximenia americana* mediated AgNPs could be an alternative source for mosquito larvicides because they constitute a potential source of bioactive chemicals and generally free from harmful effects. These results could encourage the search for new active isolated compounds offering an alternative to synthetic insecticides from other medicinal plants. This research highlighted that the green synthesis of mosquitocidal AgNP using *Ximenia americana* is an efficient and eco-friendly method against *Ae. aegypti* mosquito, and their efficacy in the field is yet to be tested for effective mosquito larval control.

## Conflict of interest statement

The authors declare that there is no conflict of interest.

## Authors' contributions

MS designed, maintained mosquito rearing house, handled experiments, synthesized AgNPs, conducted bioassays,

statistical analysis and developed the manuscript. BRN identified the mosquito species, reviewed the literature and contributed to manuscript development. The author(s) read and approved the final manuscript.

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